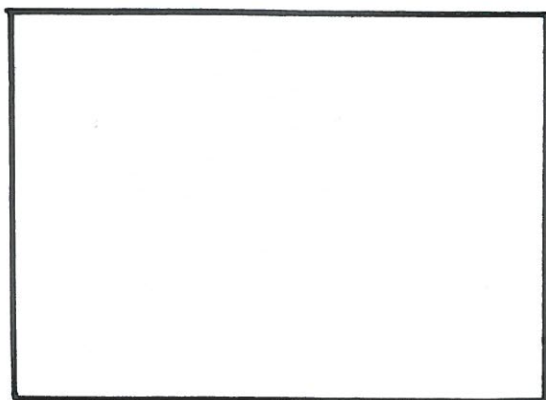




K_p - The Equilibrium Constant

Partical Pressures



☆

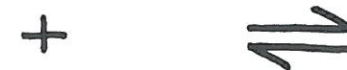
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☆

The Expression



e.g.





K_p - Numbers & Units

1 2 3 4 5

e.g. + \rightleftharpoons

$$K_p = \text{_____} = \text{_____} =$$

$$\text{UNITS} = \text{_____} = \text{_____} =$$

UNITS

e.g. + \rightleftharpoons

$$K_p = \text{_____} = \text{_____} =$$

$$\text{UNITS} = \text{_____} = \text{_____}$$

•
•
•
•
•



K_p - Deducing Partial Pressures / Mole Fractions

1 2 3 4 5

_____ were reacted with _____ in a sealed container at 500K to form _____. It was found that the _____ contained _____ at a pressure of _____



①

②

③

④

$K_p = \text{---} = \text{---} =$

Units = $\text{---} =$